

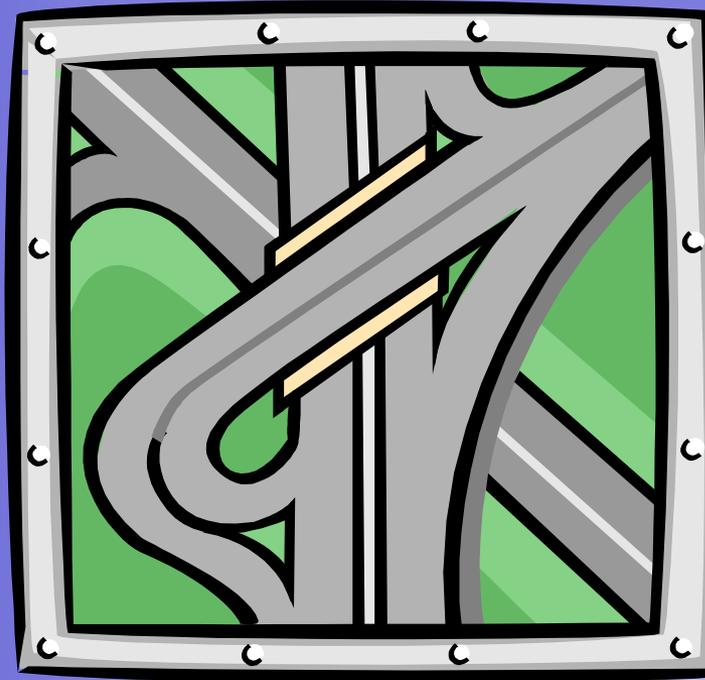
An Introduction to Groundwater Issues at Mine Sites

Produced by:

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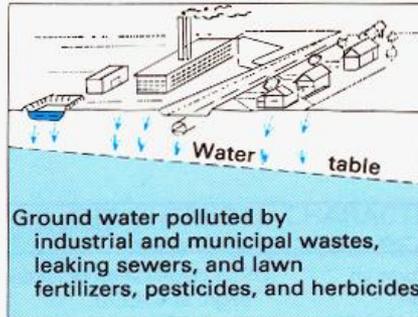


Topic 7: Contaminant Migration in Groundwater Systems

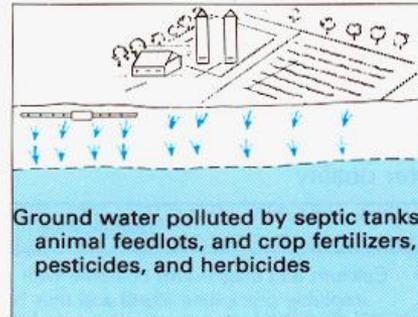


Contamination Migration in Groundwater Systems

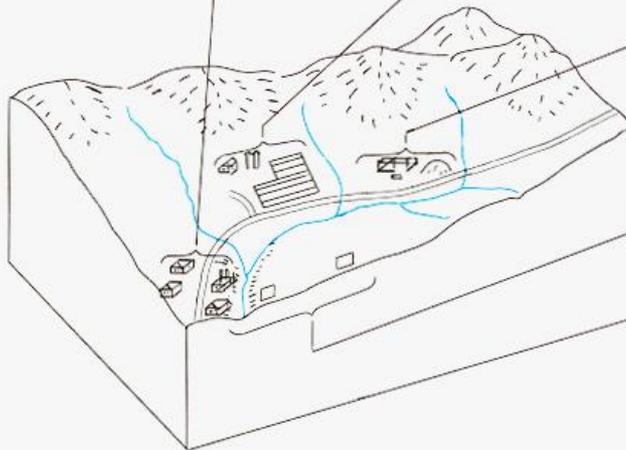
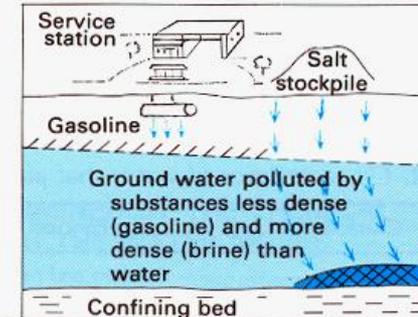
URBAN AREAS



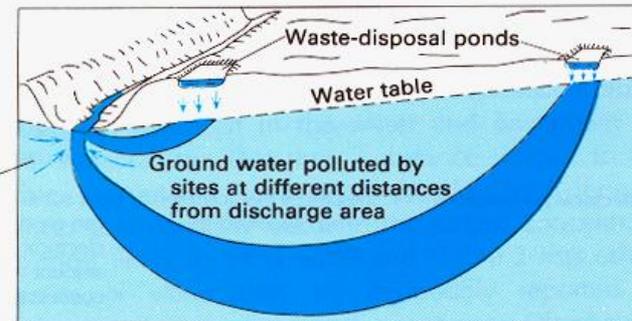
RURAL AREAS



DENSITY EFFECTS



DISTANCE EFFECTS



Example Sources of Contamination

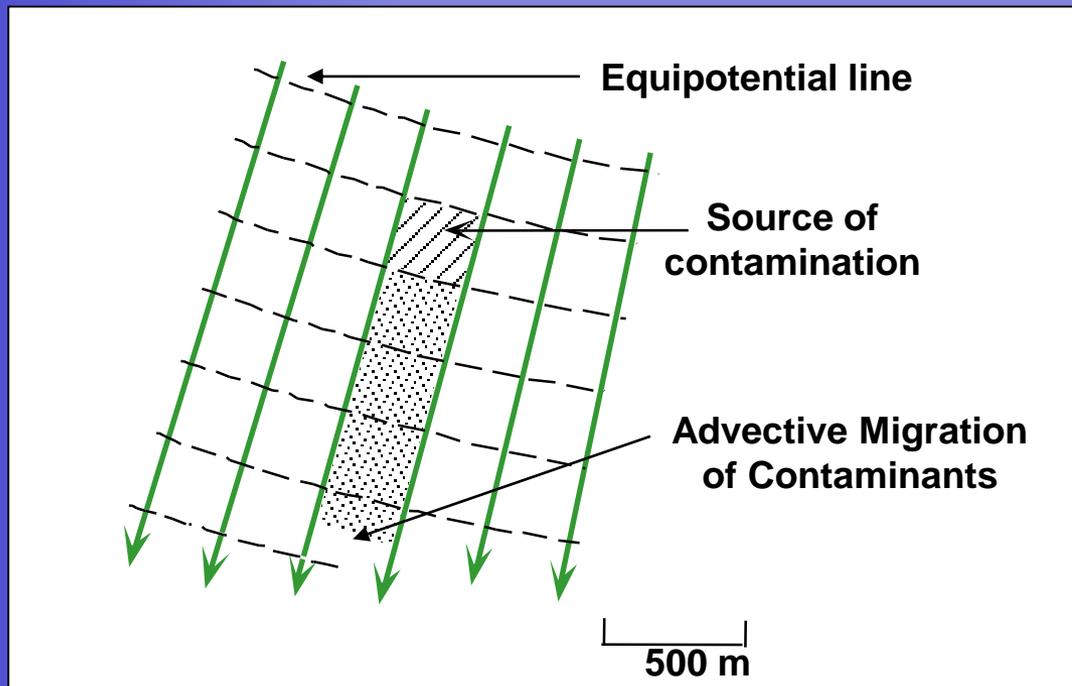
- Holding ponds (unlined / leaky)
- Tailings (with process water / other lechate)
- Waste rock (reactive – eg. acidity)
- Fuel spills
- Process chemical spills
- Smelter gases (outfall as particulate / in precipitation)

Process Contributing to Transport

- Advection
 - migration with water flow
- Dispersion
 - mixing and dilution during flow
- Diffusion
 - slow, random concentration driven migration

Advection

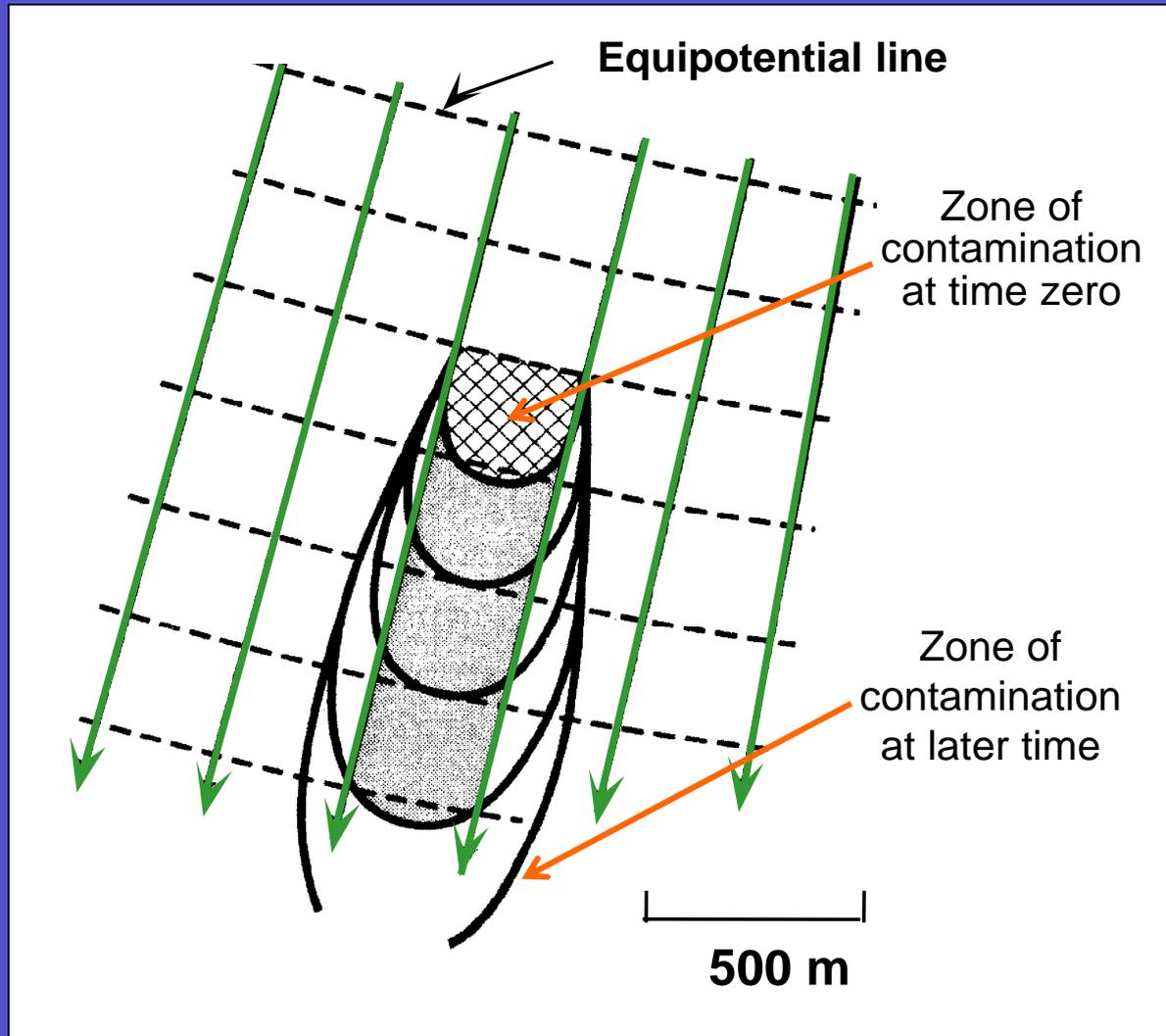
- Depends on velocity
 - $\bar{v} = q/n = k/n \, dh/dl$
 - Distance = $\bar{v} * t$



Dispersion

- Mixing at boundaries of plume
- Due to
 - Microscopic effect – flow around grains
 - Macroscopic effects – variation in flow rates in layers etc.
- Causes contaminations to travel “faster” (at more dilute concentrations) and linger longer when rehabilitation is applied

Dispersion



Diffusion

Fick's Law

- $F = -D \, dc/dx$

Where:

F is flux [$M \, L^{-2} \, T^{-1}$]

C is concentration [$M \, L^{-3}$]

x is distance [L]

D is proportionality constant called the
Diffusion Coefficient [$L^2 \, T^{-1}$]

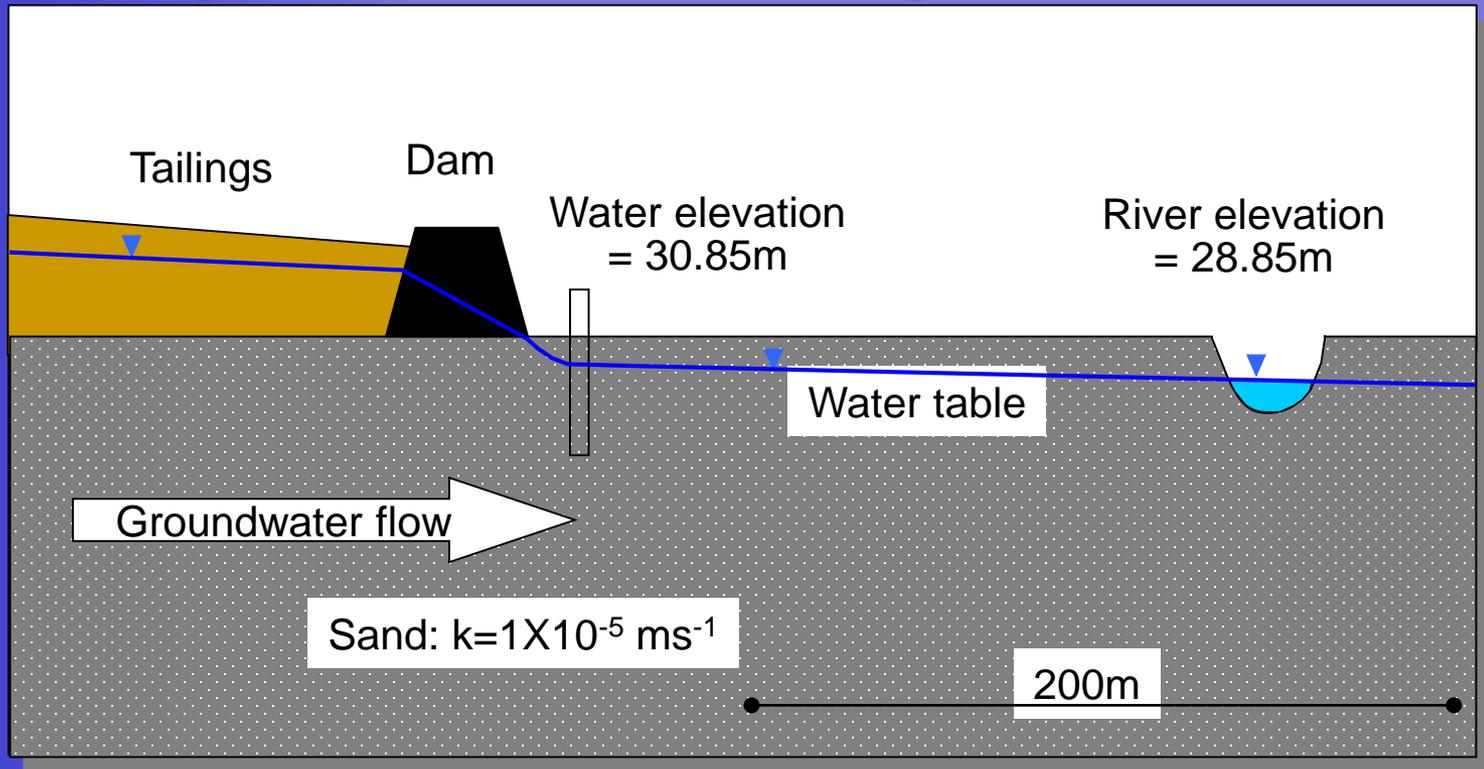


Comparing Effects of Advection, Dispersion and Diffusion

- In high K systems (sands and gravels) **advection** dominates
- In low K systems (clay) **diffusion** dominates
- At screening level calculations it is convenient to select one OR the other condition

Example: Estimating Travel Time

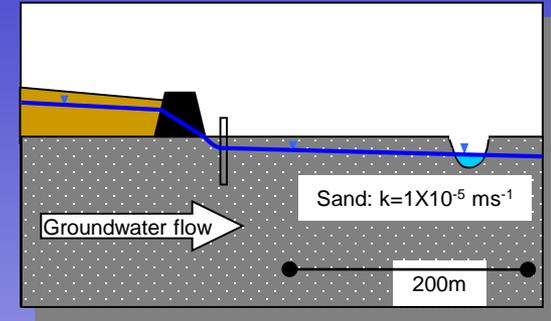
- Estimate the travel time for tailings water to migrate to a river via groundwater



Example: Estimating Travel Time

Assume

- Porosity = 0.40
- Steady flow
- Constant hydraulic gradient



Hydraulic gradient

$$dh/dl = (30.85-28.85)m/200m = 2/200 = 0.01$$

Velocity

$$V = K/n dh/dl = 1 \times 10^{-5} \text{ms}^{-1} / 0.40 * 0.01 = 2.5 \times 10^{-7} \text{ms}^{-1}$$
$$= 7.9 \text{ ma}^{-1}$$

Time

$$\text{Time} = \text{distance/velocity} = 200m / 7.9 \text{ ma}^{-1} = \sim 25a$$

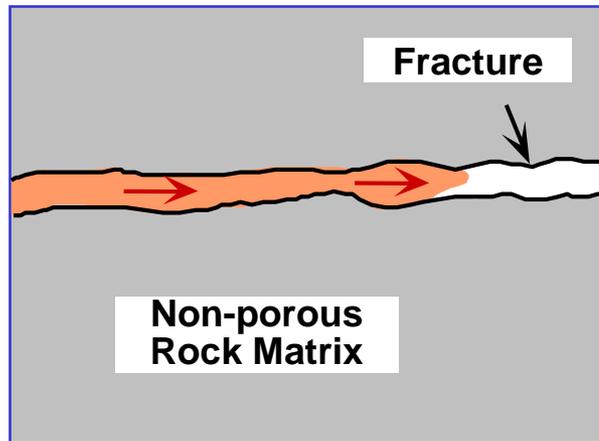
[if $k=1 \times 10^{-7}$ (silt); Time = 2,500 a for same gradient]

Flow and Transport in Fractured Rock

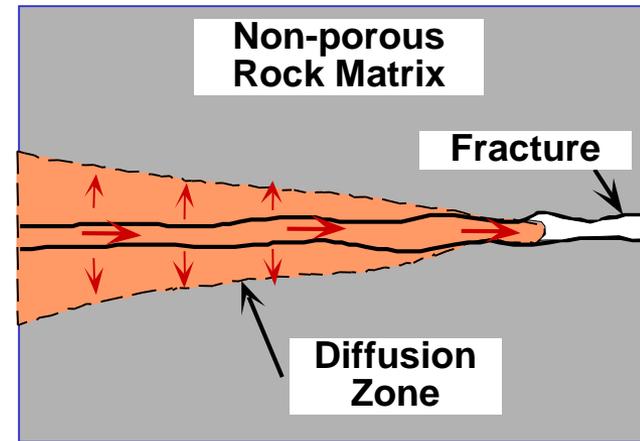
Non porous

Porous

ADVECTION IN FRACTURES

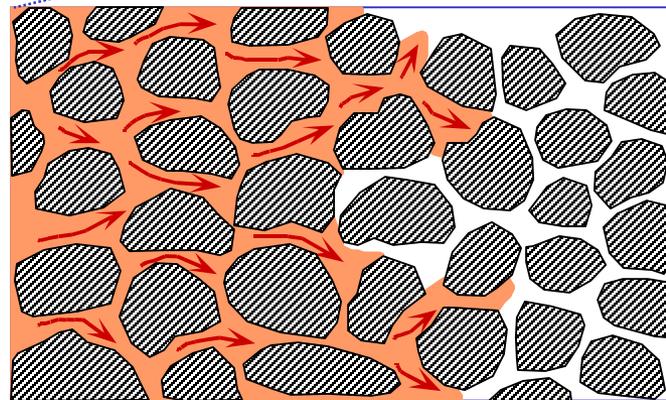
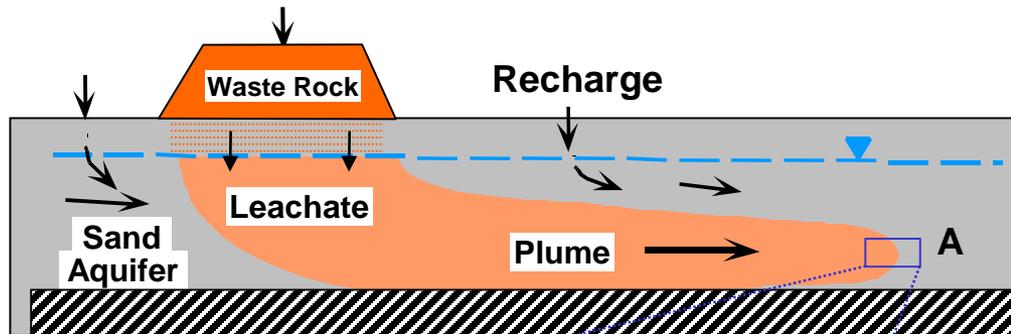


ADVECTION AND DIFFUSION



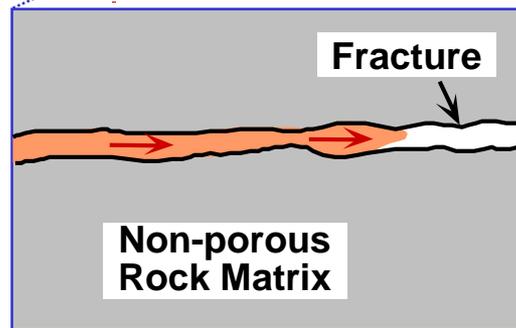
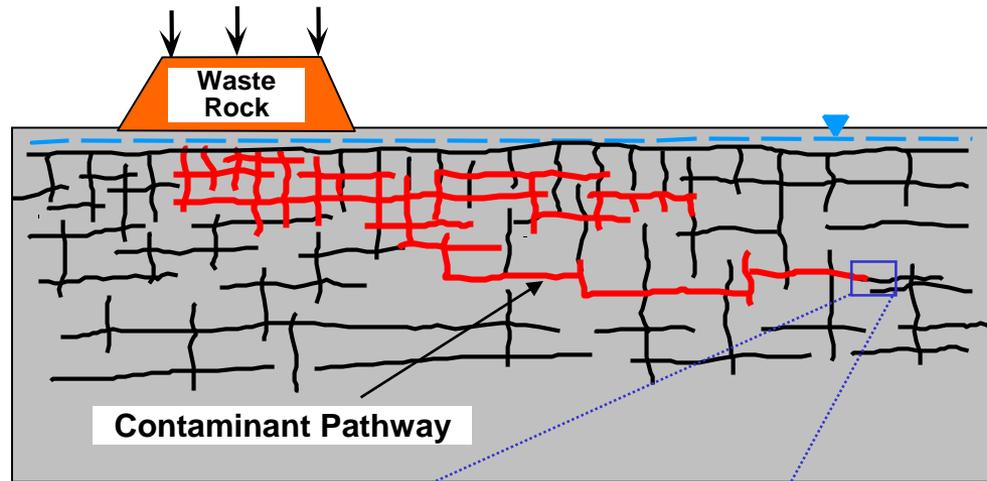
Contaminant Migration in Porous Media

CONTAMINANT MIGRATION CONTROLLED BY ADVECTION AND DISPERSION



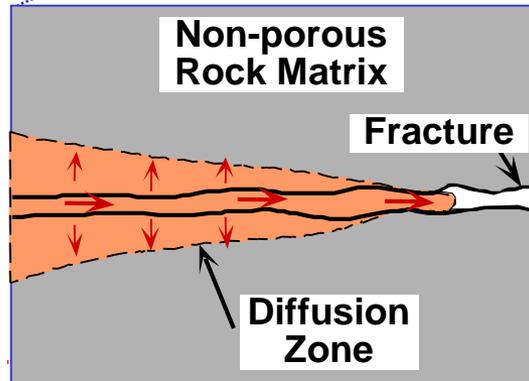
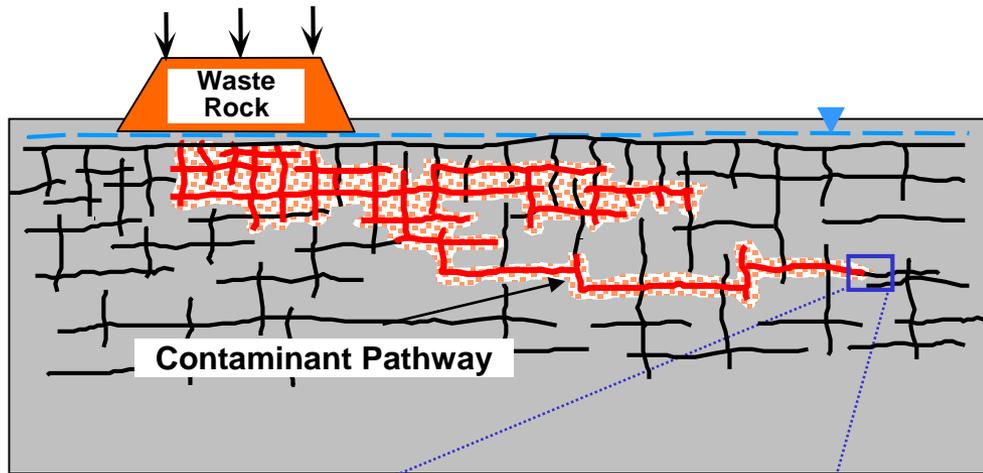
DISPERSION ZONE AT FRONT

Plume Migration in Fractured Non-Porous Rock



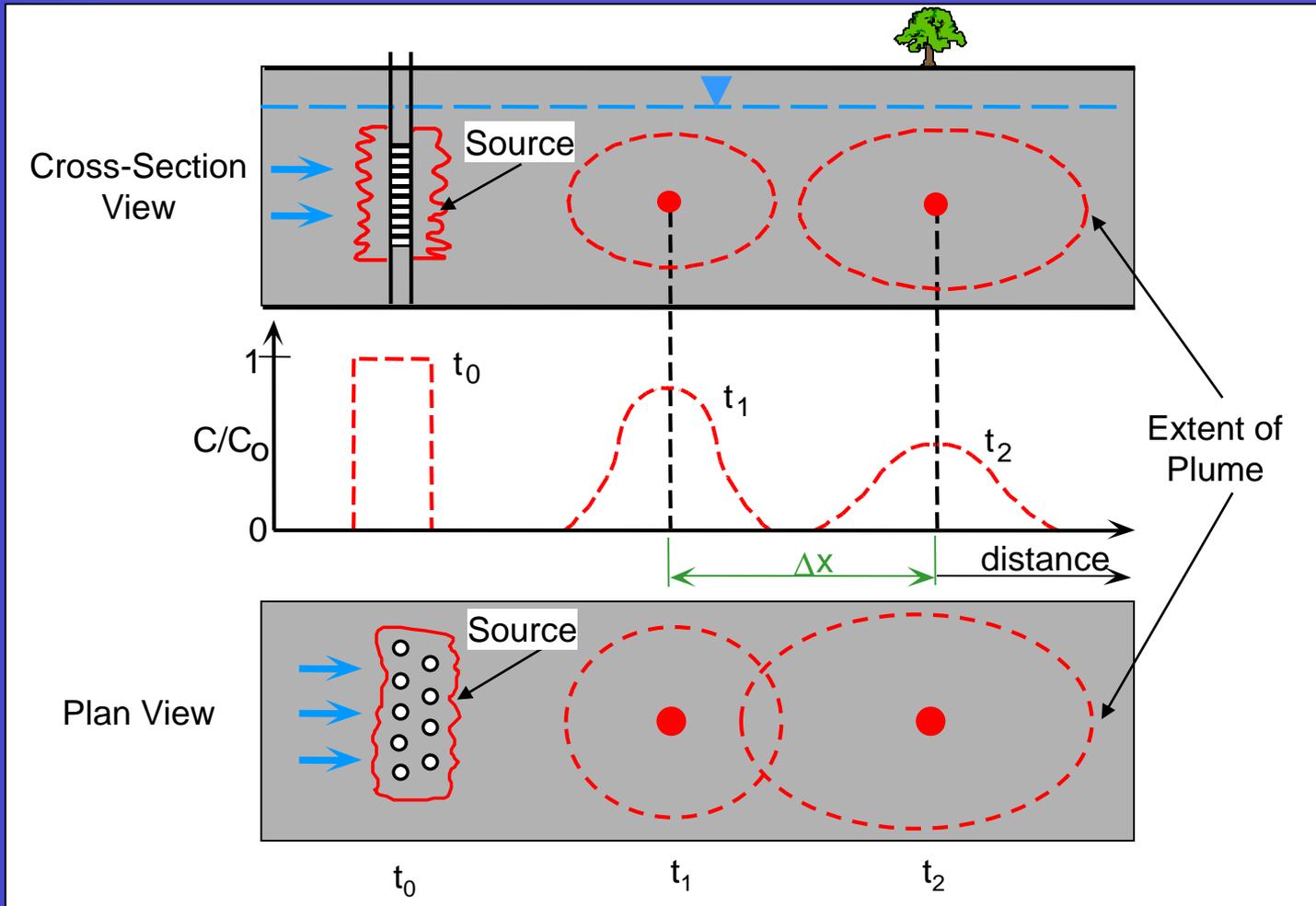
ADVECTION IN FRACTURES

Plume Migration in Fractured Non-Porous Rock



ADVECTION AND DIFFUSION

Extent of Plume

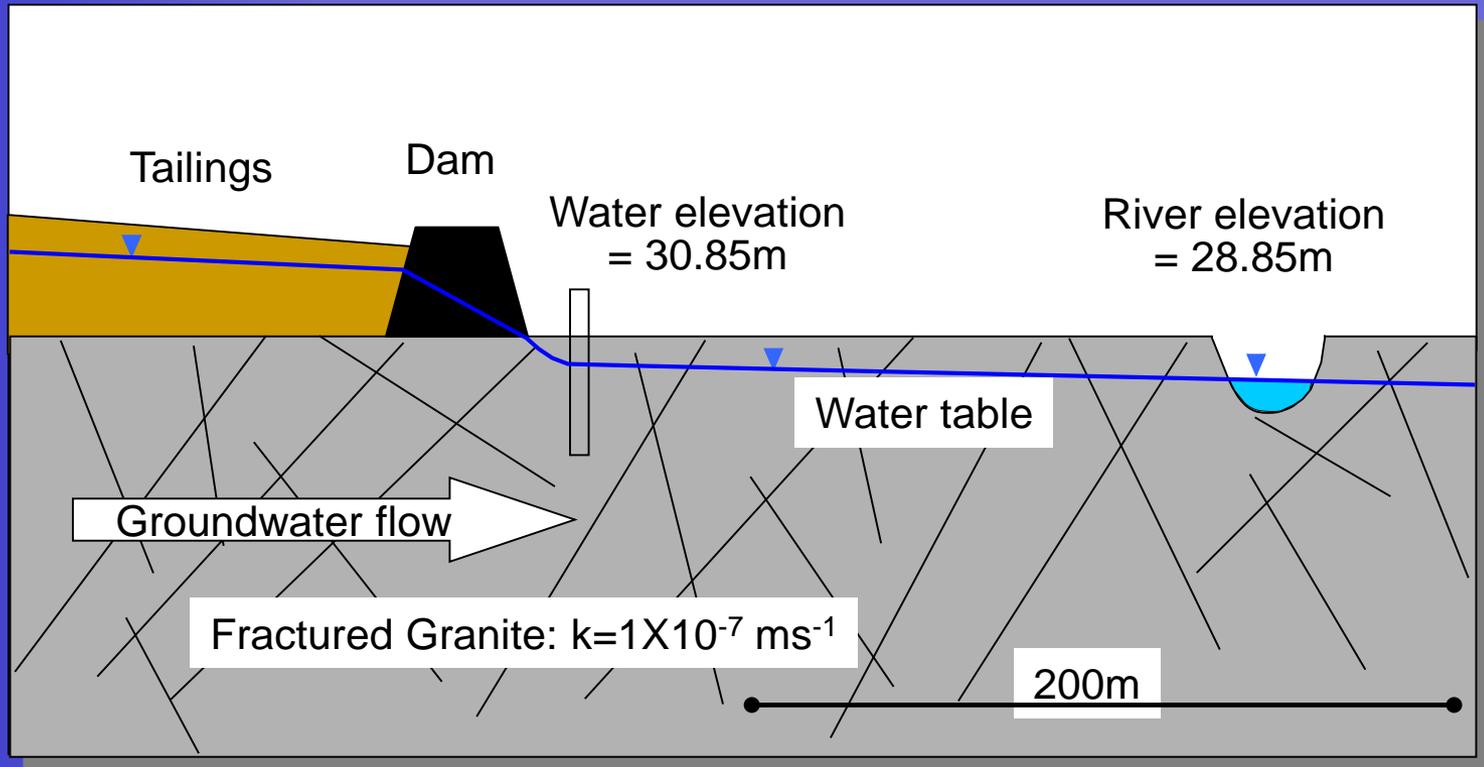


Significance of Flow Through Fractures

- May be non-uniform (follows fractures)
- Complex to monitor (hit and miss)
- Lower porosities than porous media / much higher velocities and shorter travel times



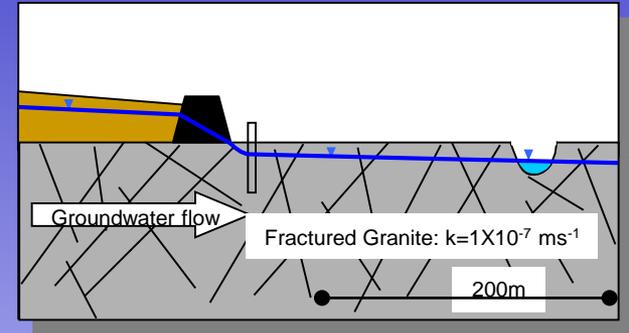
Example: Travel Time through Fractured Granite



Example: Travel Time through Fractured Granite

Assume

- Porosity = 0.004
- Steady flow
- Hydraulic gradient 0.01



Velocity

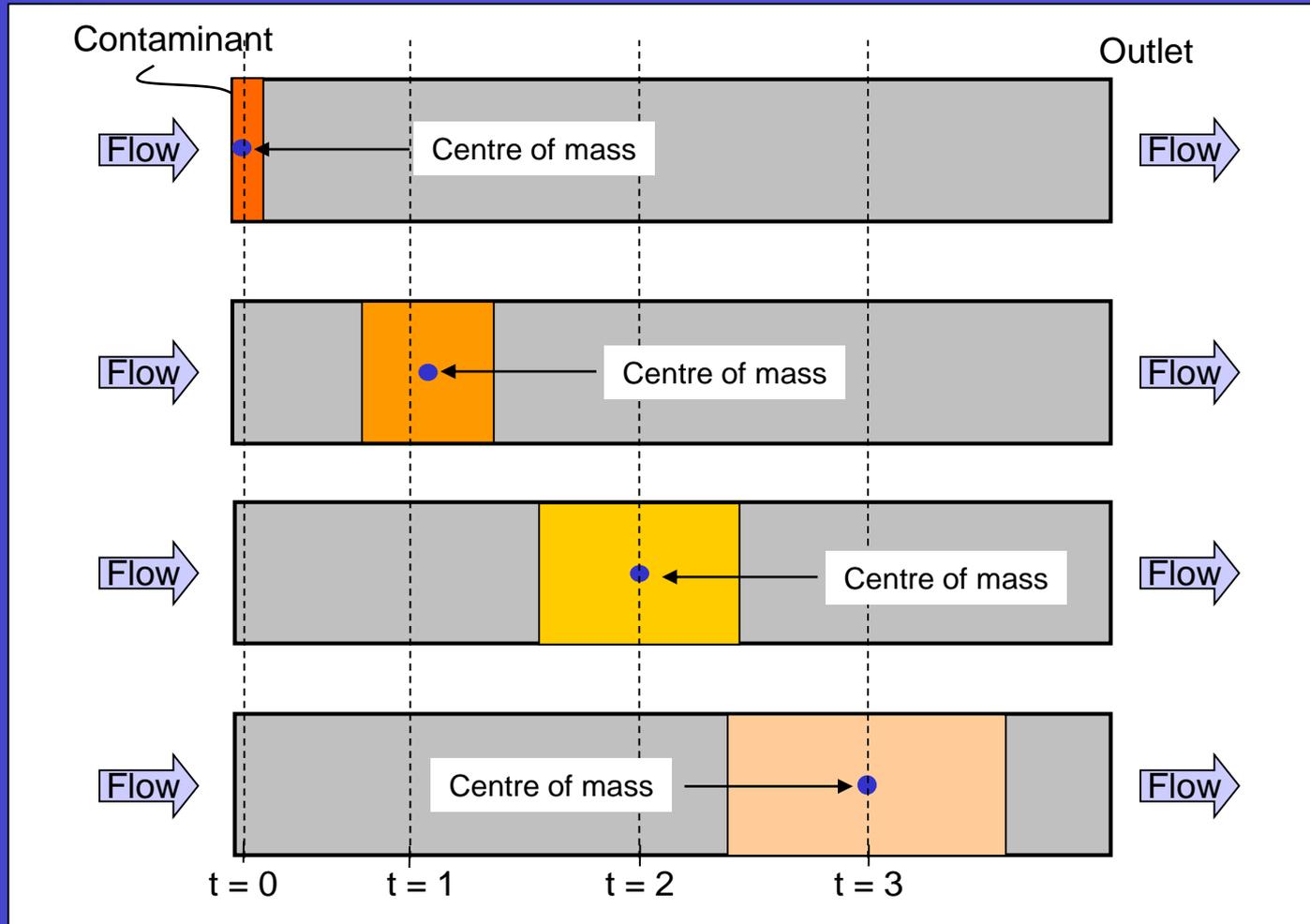
$$V = k/n \, dh/dl = 1 \times 10^{-7} \text{ ms}^{-1} / 0.004 * 0.01 = 2.5 \times 10^{-7} \text{ ms}^{-1} \\ = 7.9 \text{ ma}^{-1}$$

Time

$$\text{Time} = \text{distance/velocity} = 200\text{m} / 7.9 \text{ ma}^{-1} = \sim 25\text{a}$$

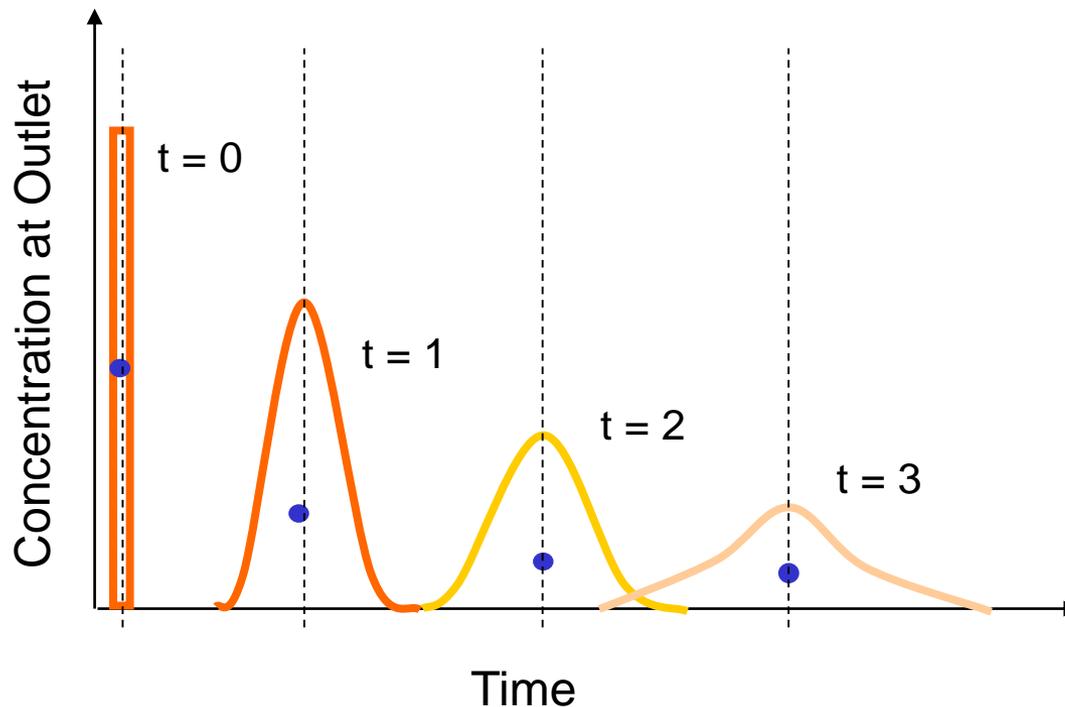
This value is same as the example with more permeable sand $k=1 \times 10^{-5} \text{ ms}^{-1}$

Effect of Dispersion on Travel Times



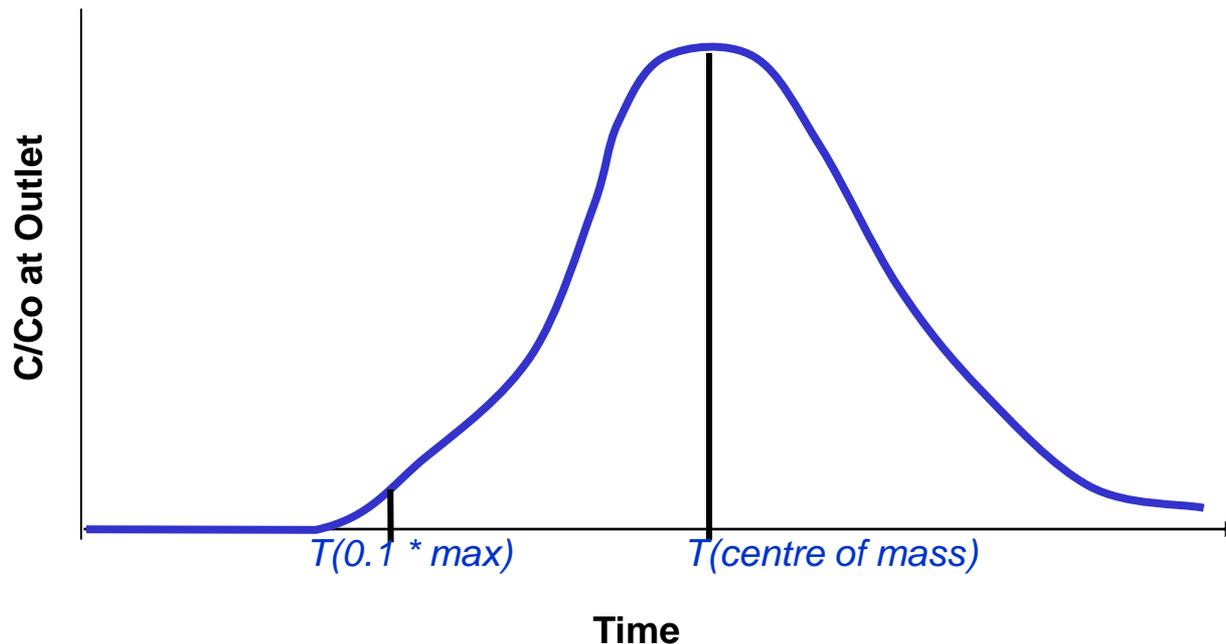
Effect of Dispersion on Travel Times

Lower concentrations arrive earlier as a result of dispersion. Important if low relative concentrations trigger regulatory or environmental limits



Effect of Dispersion on Travel Times

Lower concentrations arrive earlier as a result of dispersion. Important if low relative concentrations trigger regulatory or environmental limits

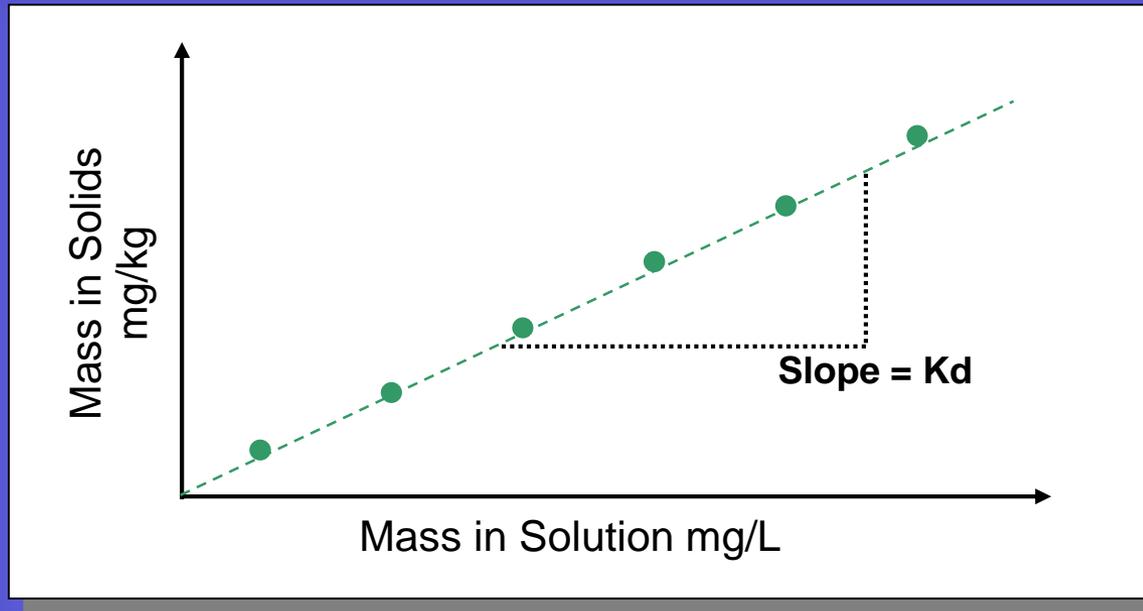


Attenuation and Retardation of Contaminants

- Many dissolved chemical constituents “react” with geologic media and thereby move slower than groundwater
- Reactions including “sorption”, ion exchange and mineral precipitation
- A simple “model” for attenuation is based on the “distribution” between water and solids is known as the **distribution coefficient (K_d)**



Distribution Coefficient (K_d)



- Good for many constituents at low concentrations
- Usually not valid over a wide range of concentrations or geochemical conditions
- units of mL/g or L/kg are common

Retardation Coefficient (R)

- $R = 1 + K_d \rho_b / n$

Where: ρ_b = bulk density (kg/L or g/cm³)

n = porosity (unitless)

K_d = distribution coefficient (mL/g or L/kg)

- Generally ρ_b is in the range of 3 to 6 kg/L for unconsolidated sediments
- So that $R \cong 1 + 4 K_d$ (so $R \geq 1$)

Velocity of Retarded Solutes (v_s)

- $v_s = v_w / R$

Where v_w is the average velocity of the groundwater

- For example, in sand if the following Kd values for arsenic (As) and Ra-226 would give:

	Kd (kg/L)	R
As	1.3	6.2
Ra-226	1000	4300

- For the groundwater travel time calculated in a previous example:

Substance	Travel Time
water	25a
As	155a
Ra-226	107,500a



Factors affecting Attenuation (Kd values)

- Metals generally have higher attenuation in organic materials (swamps, lake sediments etc.)
- Clays result in higher attenuation
- Iron oxides (rusty red-brown colour) can also cause higher attenuation
- Attenuation is higher in porous materials than in fractured rock
- Specific chemical reactions can result in high attenuation (detailed chemical modelling maybe appropriate)

Flux and Loading Rates

- **Mass Flux** = Rate of mass crossing an area per unit time [$M L^{-2} T^{-1}$]
- **Loading Rate** = Mass entering / leaving per unit time [$M T^{-1}$]
- In groundwater:
 Mass Flux = $C_i * q_i$
 where q_i is the volumetric flux [$L^3 L^{-2} T^{-1}$]
 Loading Rate = $C * Q$
 Where Q is the flow rate [$L^3 T^{-1}$]

Example: Tailings Near a River

Groundwater below tailings has a cyanide level of 100 mg/L with no attenuation/ degradation.

What is the cyanide flux to the river?

Calculate:

$$q = k \frac{dh}{dl} = 1 \times 10^{-5} \text{ m s}^{-1} = 1 \times 10^{-7} \text{ m s}^{-1} \\ = 3.15 \text{ m a}^{-1} \text{ or } 3.15 \text{ m}^3 \text{ m}^{-2} \text{ a}^{-1}$$

$$\text{If } C = 100 \text{ mg L}^{-1} = 100 \text{ g m}^{-3} = 0.1 \text{ kg m}^{-3}$$

$$\text{Then mass flux} = C * q = 0.1 \text{ kg m}^{-3} * 3.15 \text{ m}^3 \text{ m}^{-2} \text{ a}^{-1} \\ = 0.3 \text{ kg m}^{-2} \text{ a}^{-1} \text{ to the river}$$

Tailings and River Example (con't)

The contaminated groundwater zone is 10m deep and 200m wide (along the dam)

What is total loading of cyanide to the river?

Calculate:

$$\begin{aligned}q &= Q * A \\ &= 3.15 \text{ m}^3 \text{ m}^{-2} \text{ a}^{-1} * (200 * 10) \text{ m}^2 \\ &= 6300 \text{ m}^3 \text{ a}^{-1}\end{aligned}$$

$$\begin{aligned}\text{Mass load} &= Q * C \\ &= 6300 \text{ m}^3 \text{ a}^{-1} * 0.1 \text{ kg m}^{-3} \\ &= 630 \text{ kg a}^{-1} \text{ to the river}\end{aligned}$$



Example: Considering Dilution

What flow rate in the river is necessary to assimilate the average annual cyanide load if the concentration in the river must not exceed 1mg/L ? (Assume that cyanide upstream is negligible).

Calculate:

Mass load in river = 630 kg a⁻¹ to the river

If C = 1 mg L⁻¹ = 1 g m⁻³ = 0.001 kg m⁻³

Then Q = Load / C

$$= 630 \text{ kg a}^{-1} / 0.001 \text{ kg m}^{-3}$$

$$= 630,000 \text{ m}^3 \text{ a}^{-1}$$

$$= 1726 \text{ m}^3 \text{ day}^{-1}$$

$$= 20 \text{ L s}^{-1}$$

